

Concentration

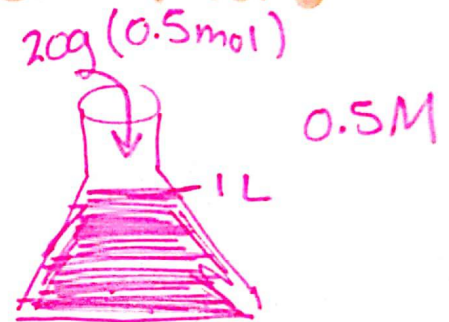
- Amount of solute dissolved expressed 2 ways:

1) Molarity (M)

↳ moles solute per liters of solutions

Volume

$$M = \frac{\text{mol}}{\text{L}}$$



2) molality (m)

↳ moles of solute per Kg of solvent

mass

$$m = \frac{\text{mol}}{\text{Kg}}$$

