Topics Covered: Physical and Chemical Changes ■Physical and Chemical **Properties** □Types of Matter Naming

UNIT OBJECTIVES

- ☐ Understand physical and chemical changes and the conservation of mass
- Understand matter in terms of definition, states, changes in state, and chemical and physical properties
- Name inorganic ionic and molecular compounds

PHYSICAL PROCES

Law of Conservation of Mass:

*Applies to physical and chemical changes

Physical Change:

Chemical Change:

Types of Chemical Changes

1. Formation of a Precipitate

Types of Chemical Changes

2. Color Change

*Dilution is NOT a chemical change

Types of Chemical Changes

3. Gas Production

Types of Chemical Changes

- 4. Heat Exchange
- Exothermic:
- Endothermic:

PHYSICACHIES PHYSICACHIES PROPERTIES

Physical Properties of Pure Substances

- 1. Electrical Conductivity
- 2. Heat Conductivity
- 3. Melting Point
- 4. Boiling Point

Physical Properties of Pure Substances

- 5. Density
- 6. Index of Refraction
- 7. Malleability
- 8. Ductility

Chemical Properties of Pure Substances

- 1. Reaction with acids and bases
- 2. Reaction with oxygen (combustion)
- 3. Acting as oxidizing/reducing agent
- 4. Reaction with other elements

Chemical Properties of Pure Substances

- 5. Decomposition into simpler substances
- 6. Corrosion

Intensive Property:

Extensive Property: