PHYSICAL V	S. CHEMICAL
PROPERTIES	

Name	
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A physical property is observed with the senses and can be determined without destroying the object. For example, color, shape, mass, length and odor are all examples of physical properties.

A chemical property indicates how a substance reacts with something else. The original substance is fundamentally changed in observing a chemical property. For example, the ability of iron to rust is a chemical property. The iron has reacted with oxygen, and the original iron metal is changed. It now exists as iron oxide, a different substance.

Classify the following properties as either chemical or physical by putting a check in the appropriate column.

	Physical Property	
1. blue color		
2. density		
3. flammability	Medican trade and a case on a specimen of the superior and development of the first of the superior and a case	
4. solubility	foret men de de menor passa de la mode de pre de mente de présentado en la serio como en ficialmentado de de c	
5. reacts with acid to form H ₂	Section metror in makes 44,000 of 4,000 decimal provincia decimal deci	
6. supports combustion		
7. sour taste		
8. melting point	that a chairm for the chairman and the c	
9. reacts with water to form a gas	CONTRACTOR AND	Mellom code and month of the code and an electrical solution consistence of the design of the code and code as a code of the c
10. reacts with a base to form water		
11. hardness		
12. boiling point		
13. can neutralize a base		
14. luster		TO control constitute control
15. odor		