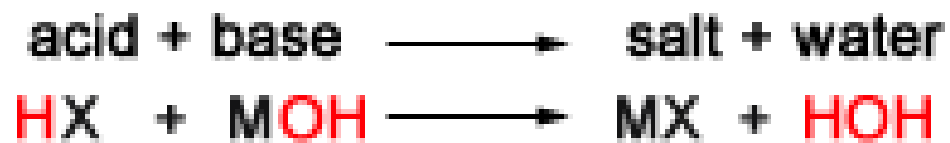


Driving Forces Notes

Acid/Base (Neutralization)

- **The driving force** of these chemical changes is **proton (H⁺) transfer**.
- The **reactants** in these reactions will lose or gain protons (hydrogen ions) and form water and a salt as products.



[Video: 08 Acid-Base Reactions](#)

Oxidation/Reduction (REDOX)

- **The driving force** of these chemical changes is **electron transfer**.
- The **reactants** in these reactions will lose or gain electrons and change their charge as they form the products.
- Easy ways to remember

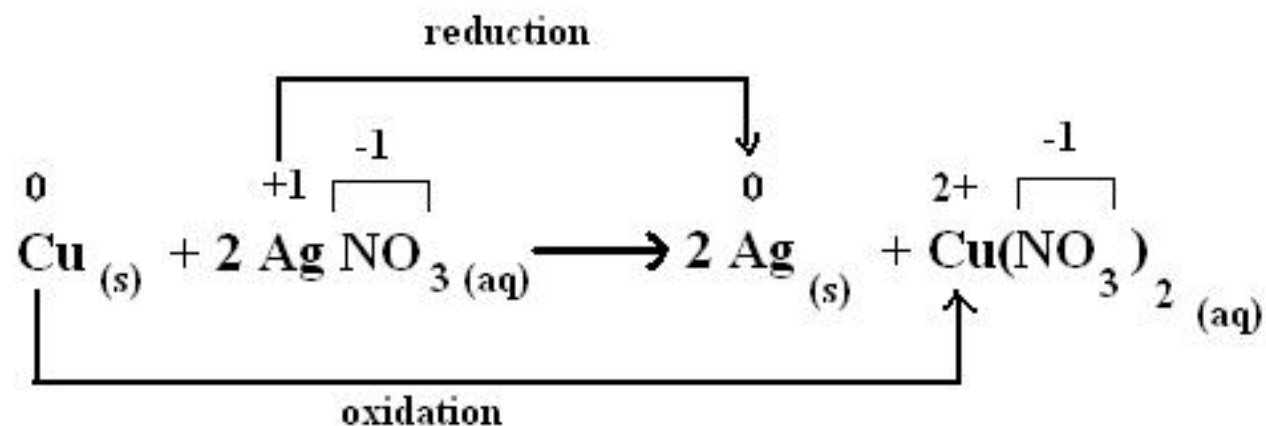
OIL RIG

Oxidation is loss (of e-),
Reduction is gain (of e-)



LEO the lion goes GER

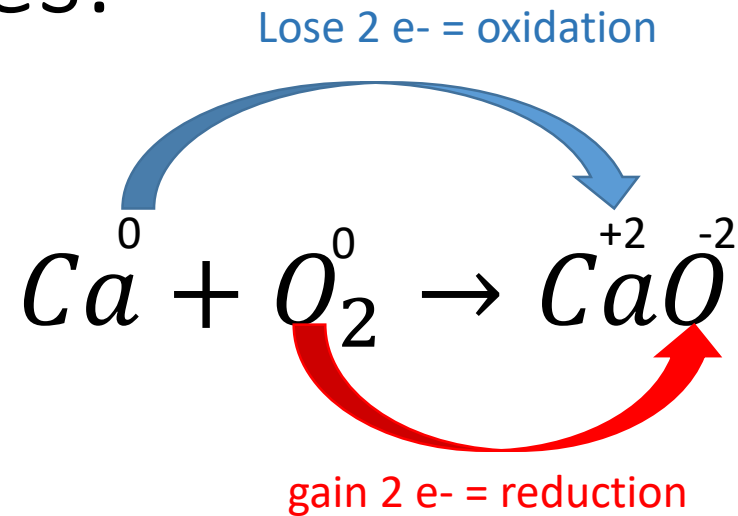
Lose electrons = oxidation,
Gain electrons = reduction



[Video: Introduction to Oxidation Reduction \(Redox\) Reactions](#)

REDOX Examples:

- Example 1:



- Example 2

