Naming Ionic Compounds

To name a compound that contains two elements,

- identify the cation and anion.
- name the cation first followed by the name of the anion.

Guide to Naming Ionic Compounds with Metals That Form a Single Ion

> STEP 1 Identify the cation and anion

STEP 2 Name the cation by its element name.

STEP 3

Name the anion by changing the last part of its element name to ide.

STEP 4
Write the name of the cation first and the name of the anion second.

To name a compound that contains polyatomic ion(s),

- identify the cation and anion.
- name the cation first followed by the name of the anion.

use polayatomic ion chart

Examples of Ionic Compounds

Formula	lons		Name
	cation	n anion	
NaCl	Na⁺	CI-	sodium chloride
K_2S	K ⁺	S ²⁻	potassium sulfide
$MgCO_3$	Mg^{2+}	CO ₃ ²⁻	magnesium carbonate
Ca(NO ₂)	Ca ²⁺	NO ₂ -	calcium nitrate
Al_2O_3	Al ³⁺	O ²⁻	aluminum oxide